

Figure 5-17 Microbiology, 7/e
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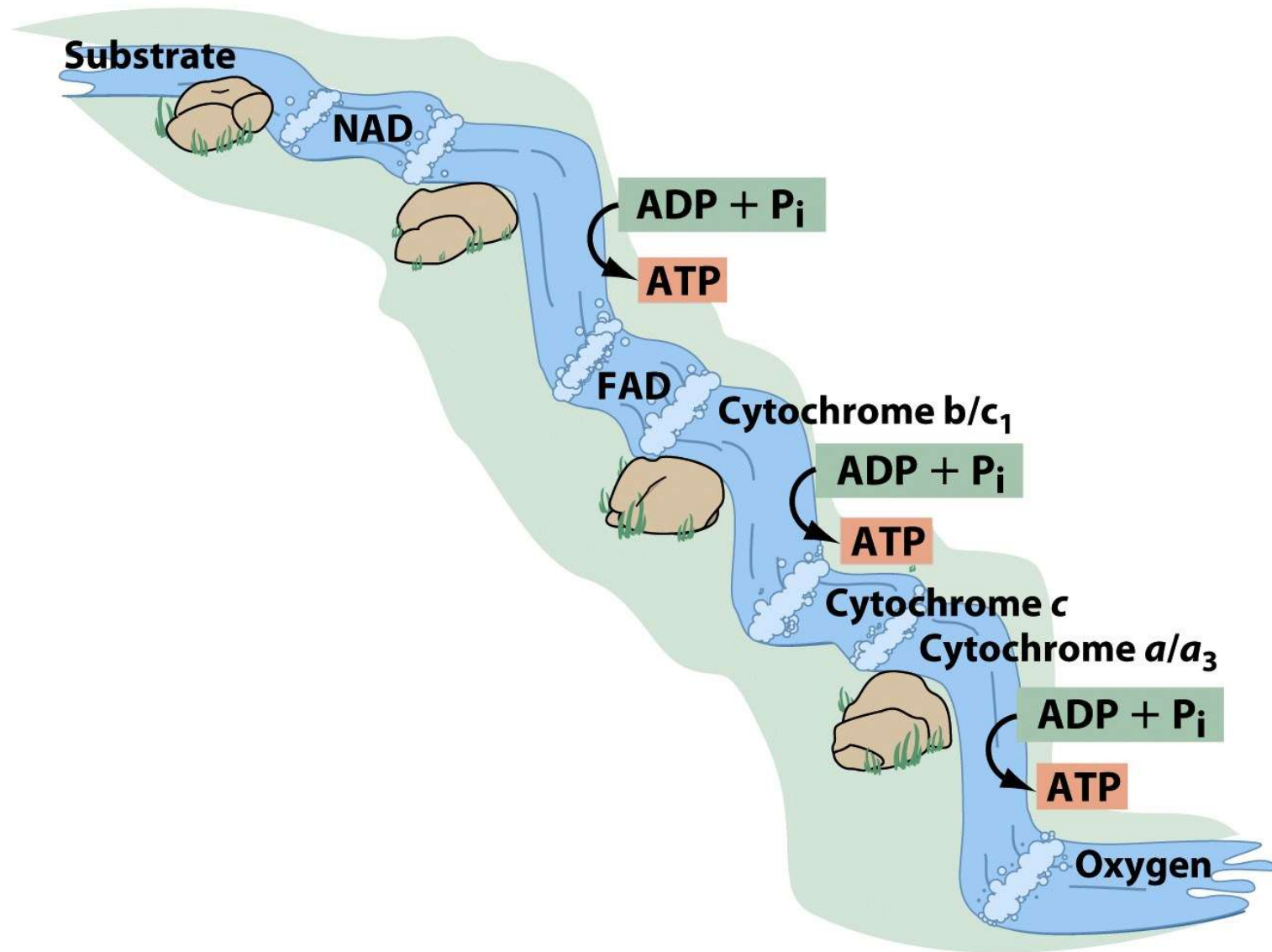
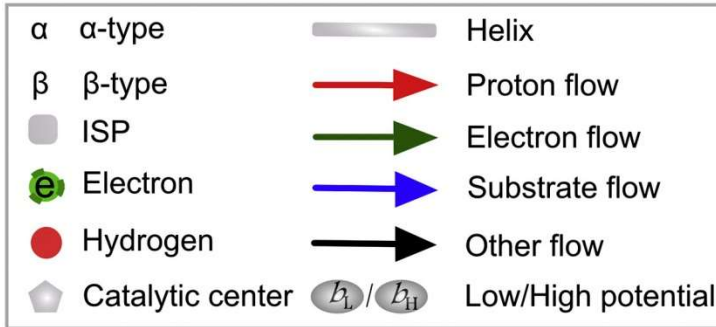
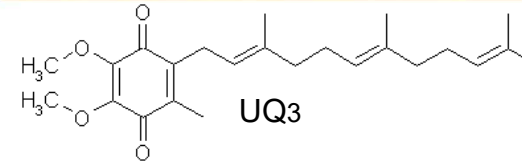


Figure 5-18 Microbiology, 7/e
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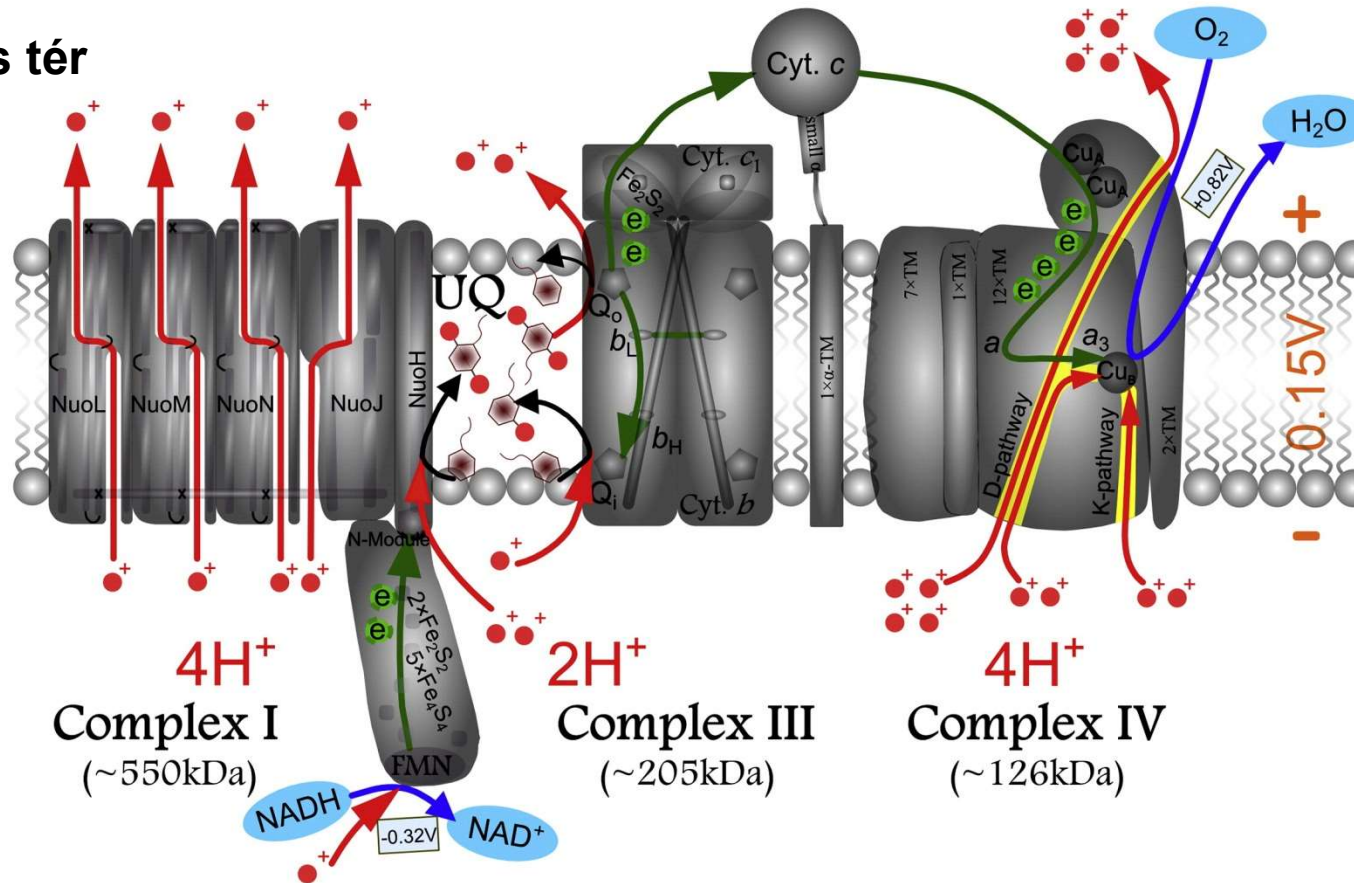


10 protons are pumped per $2e^-$

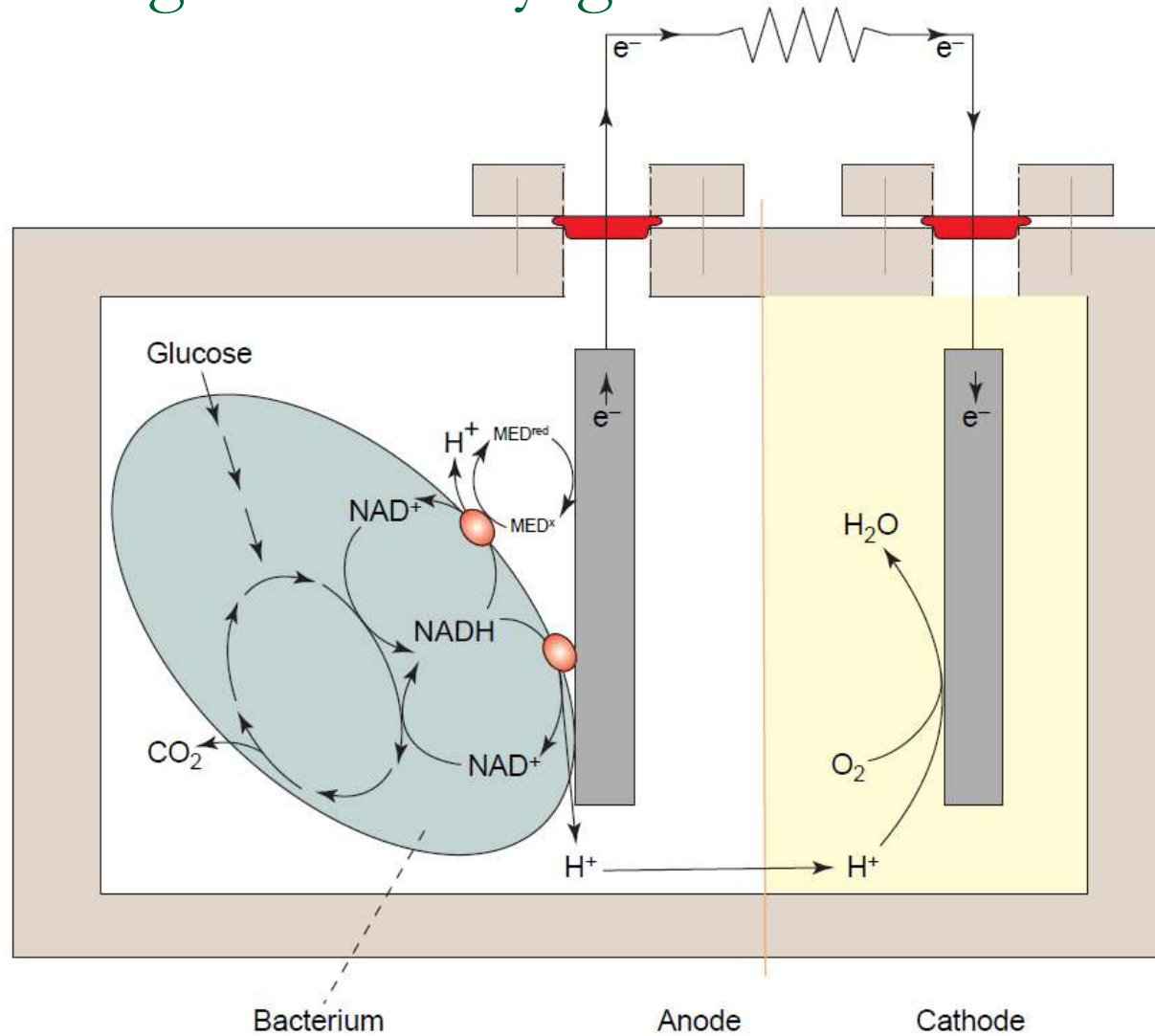


Periplazmás tér

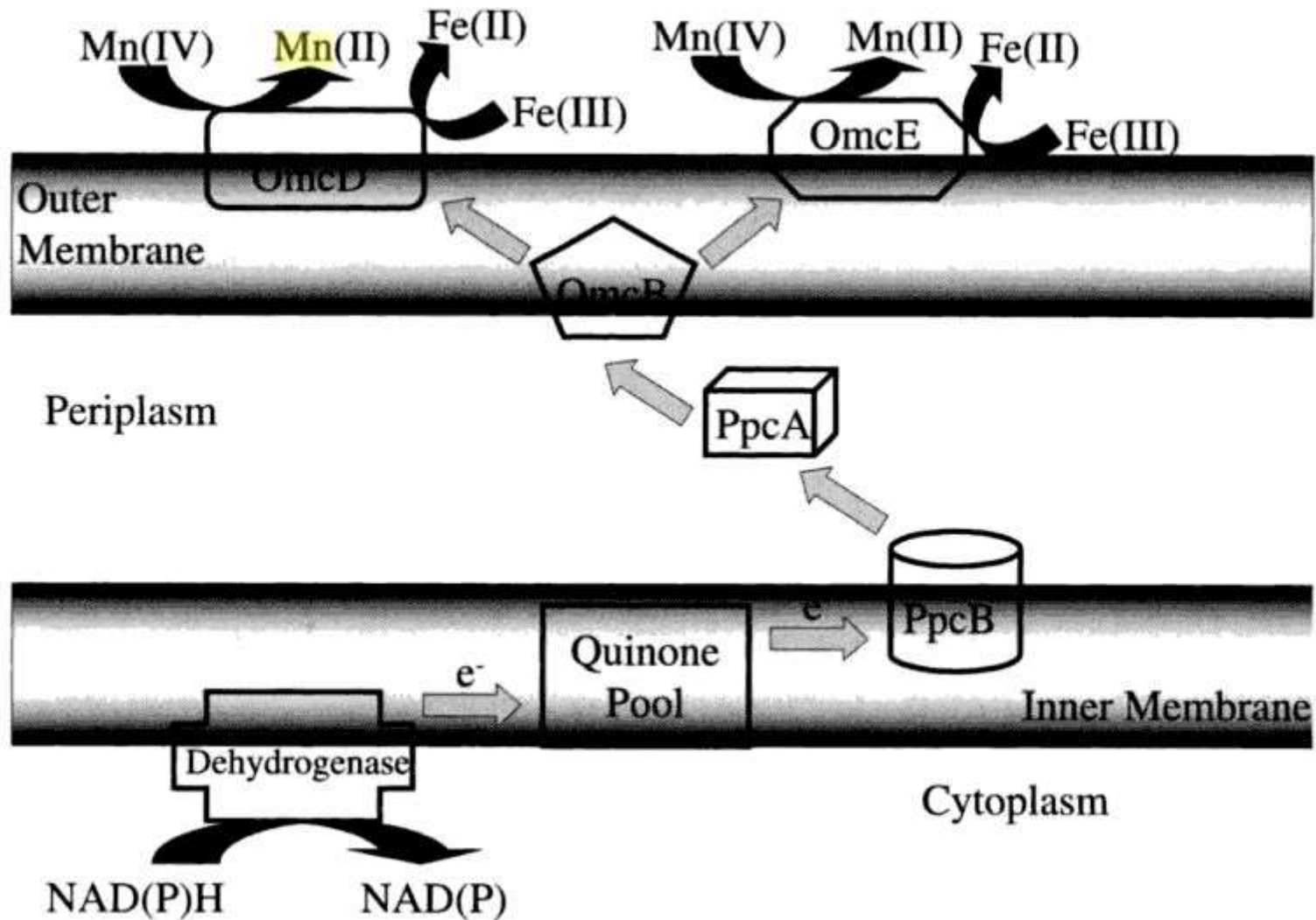
Citoplazma



Mikrobiológiai üzemanyag cella

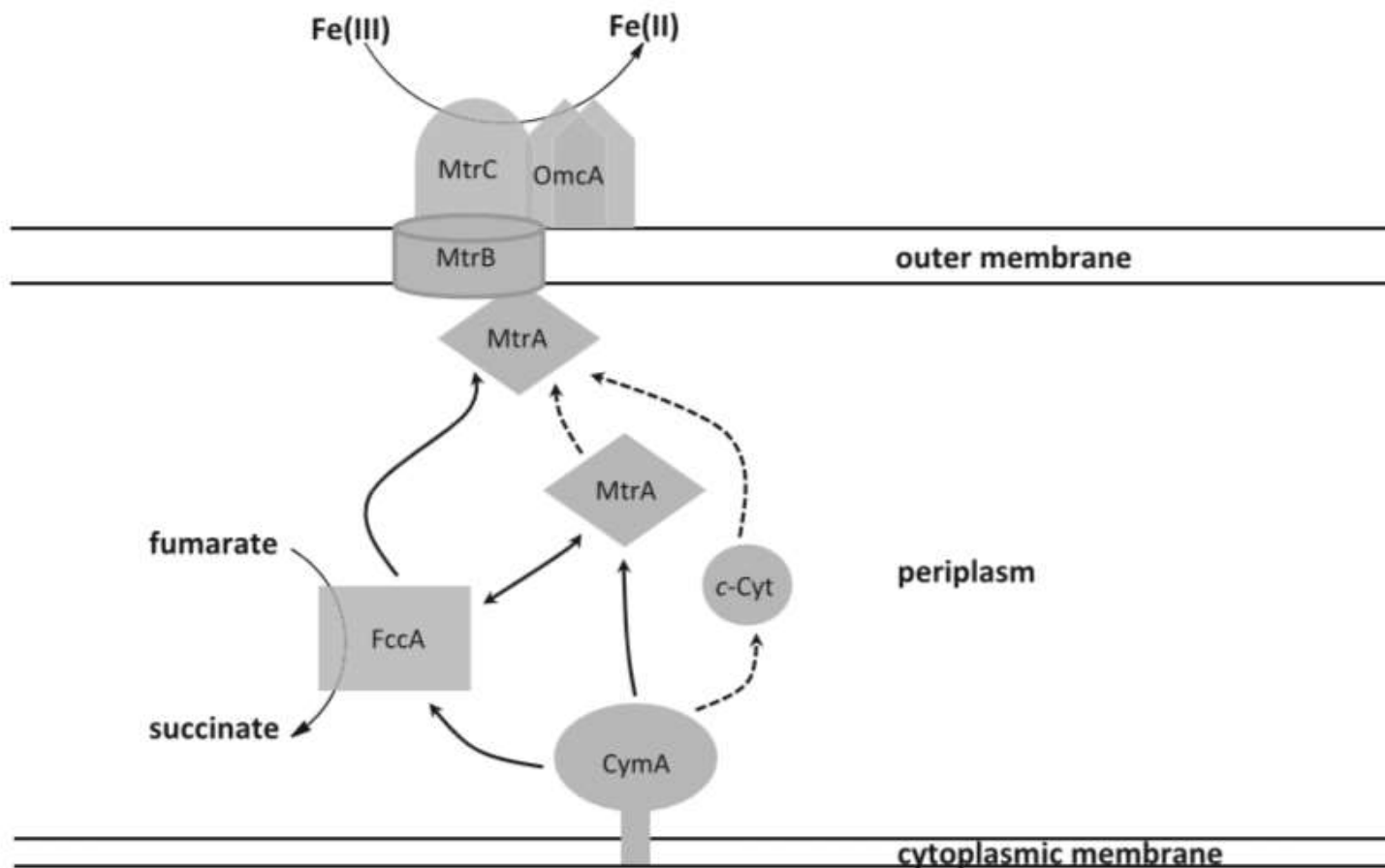


Elektron transzport lánc mechanizmusa a *Geobacter* nemzetség tagjainál

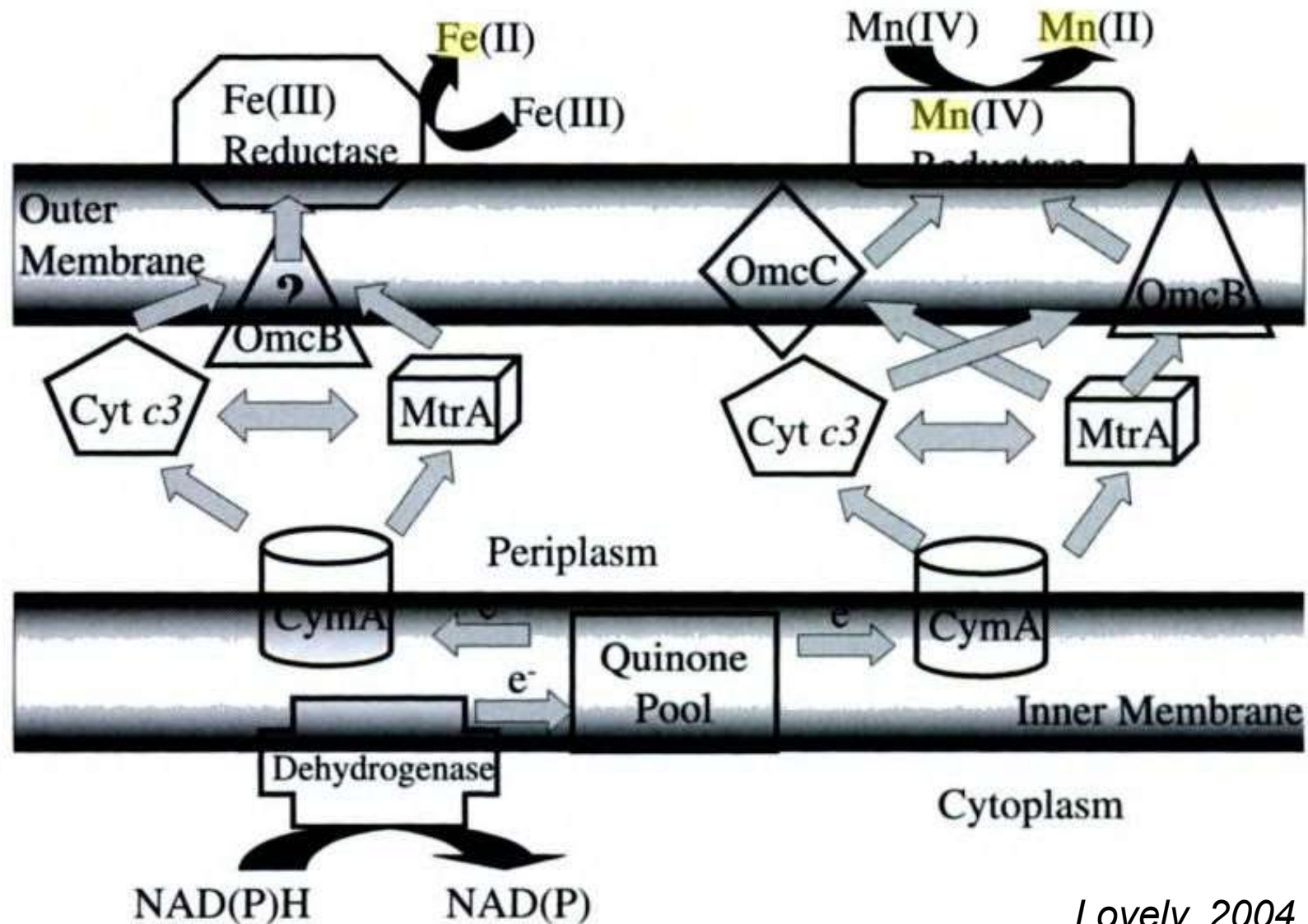


Lovely, 2004

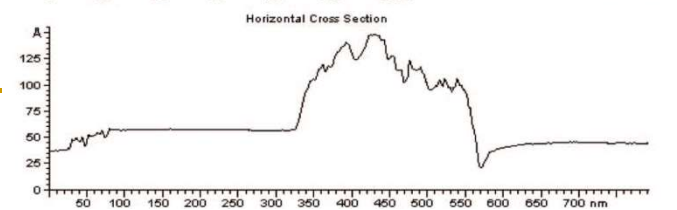
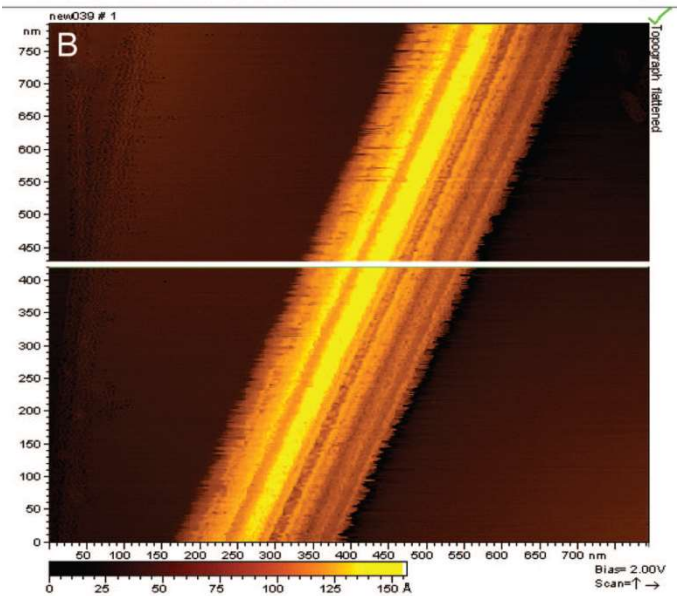
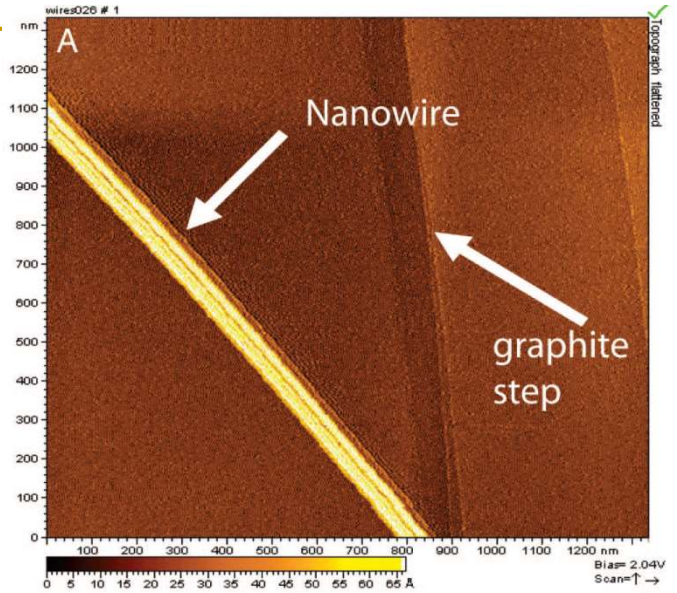
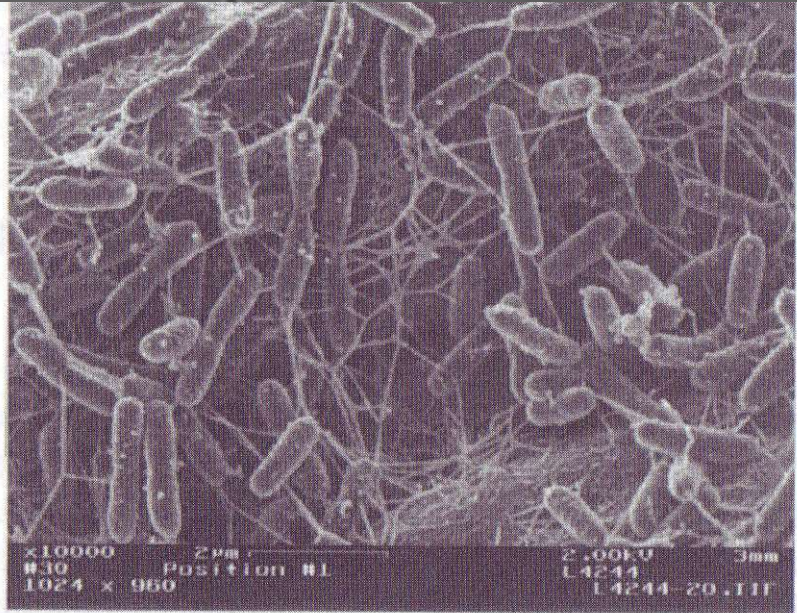
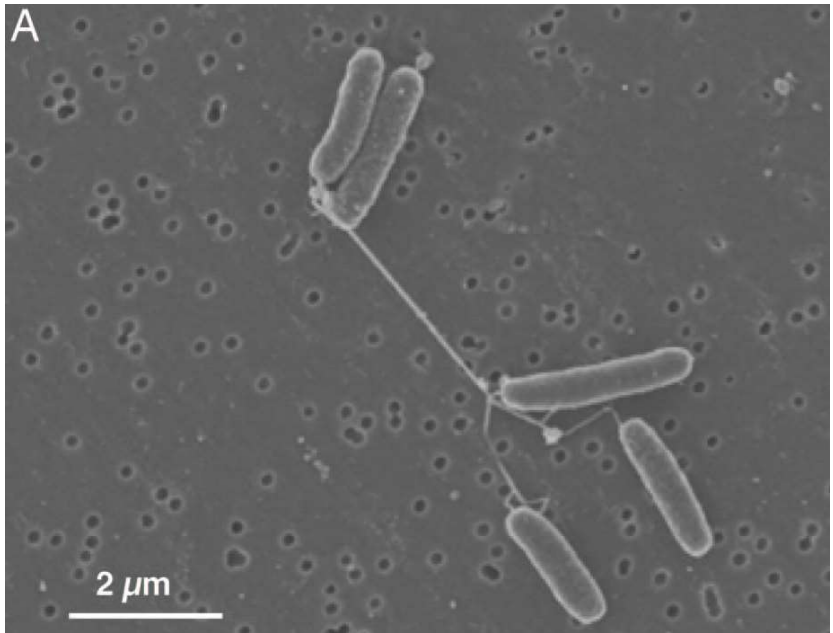
Elektron transzport lánc mechanizmusa a Shewanella nemzetség tagjainál



Elektron transzport lánc mechanizmusa a *Shewanella* nemzetség tagjainál



Lovely, 2004



Gorby et al. (2006) PNAS

Új lap | Shewanella oneidensis MR-1 | 12883.full.pdf | www.pnas.org/content/pnas/111/35/12883.full.pdf | Keresés

4 összesen: 6 | 200%

observation of the target nanowires in the platform (Movies S7 and S8). We observed localization at the periphery of the cell, as observed clear localization of these cyto-membrane-stained bacterial nanowires (25 of led with anti-MtrC and 19 of 22 nanowires OmcA), whereas no fluorescence was detected in negative control cells or their membrane of 22 nanowires labeled with anti-MtrC and res labeled with anti-OmcA; Fig. 3D).

ductance of *Shewanella* nanowires was pre-onstrated under nonphysiological conditions reported here are consistent with membrane ould function as nanowires to mediate EET alization of MtrC and OmcA to these mem- rovides the most compelling evidence to date, orpts the proposed multistep redox hopping (7, 18), allowing long-range electron transport network of heme cofactors that line *Shewa-* Fig. 4). We have shown that *S. oneidensis* (Fig. 3D); therefore, it is also

Fig. 4. Proposed structural model for *Shewanella* nanowires. *S. oneidensis*

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Shewanella oneidensis MR-1 nanowires are outer membrane and periplasmic extensions of the extracellular electron transport components

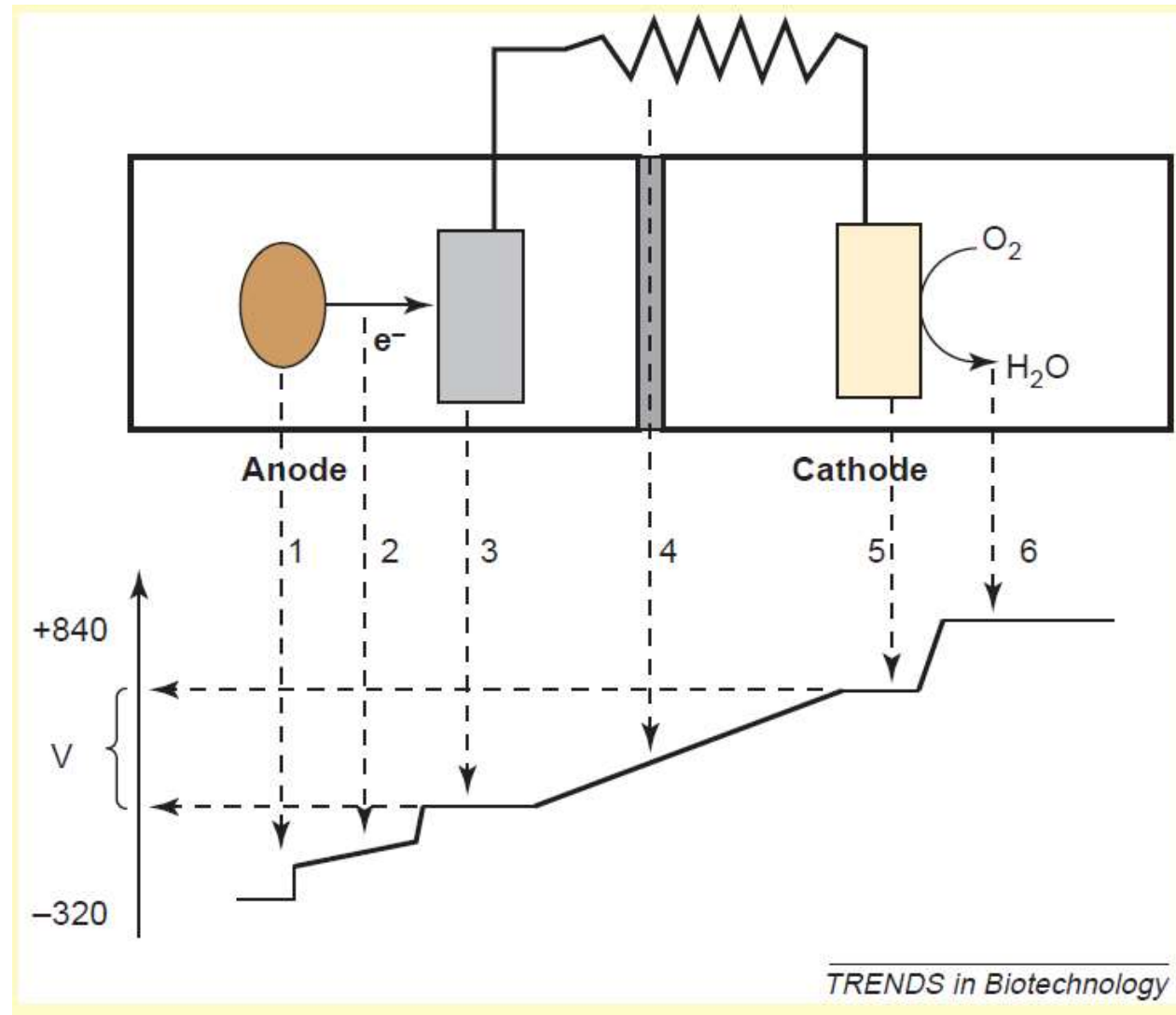
S Pirbadian, SE Barchinger... - Proceedings of the ..., 2014 - National Acad Sciences

Reakciók redoxpotenciálja

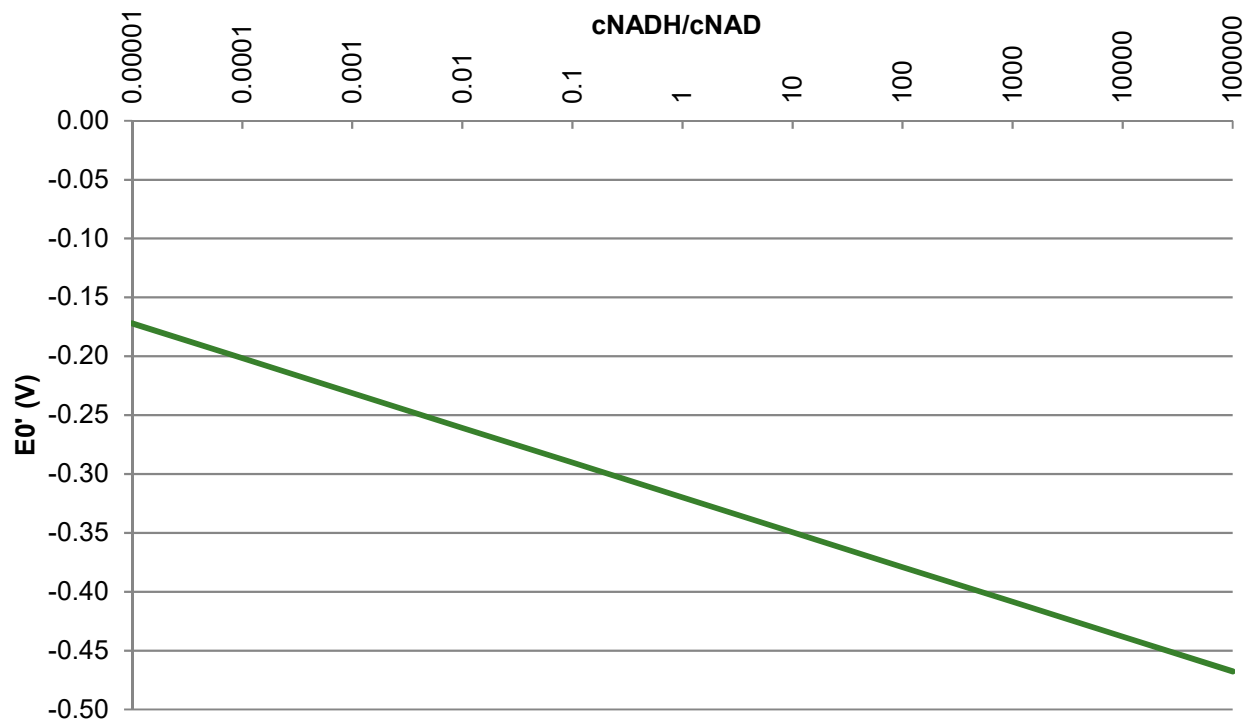
Table I

Redox reaction	E'₀ (mV)
$2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2$	-420
Ferredoxin(Fe^{3+}) + $\text{e}^- \rightarrow$ Ferredoxin(Fe^{2+})	-420
$\text{NAD}^+ + \text{H}^+ + 2\text{e}^- \rightarrow \text{NADH}$	-320
$\text{S} + 2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2\text{S}$	-274
$\text{SO}_4^{2-} + 10\text{H}^+ + 8\text{e}^- \rightarrow \text{H}_2\text{S} + 4\text{H}_2\text{O}$	-220
Pyruvate ²⁻ + $2\text{H}^+ + 2\text{e}^- \rightarrow$ Lactate ²⁻	-185
$\text{FAD} + 2\text{H}^+ + 2\text{e}^- \rightarrow \text{FADH}_2$	-180
Fumarate ²⁻ + $2\text{H}^+ + 2\text{e}^- \rightarrow$ Succinate ²⁻	+31
Cytochrome <i>b</i> (Fe^{3+}) + $\text{e}^- \rightarrow$ Cytochrome <i>b</i> (Fe^{2+})	+75
Ubiquinone + $2\text{H}^+ + 2\text{e}^- \rightarrow$ UbiquinoneH ₂	+100
Cytochrome <i>c</i> (Fe^{3+}) + $\text{e}^- \rightarrow$ Cytochrome <i>c</i> (Fe^{2+})	+254
$\text{NO}_3^- + 2\text{H}^+ + 2\text{e}^- \rightarrow \text{NO}_2^- + \text{H}_2\text{O}$	+421
$\text{NO}_2^- + 8\text{H}^+ + 6\text{e}^- \rightarrow \text{NH}_4^+ + 2\text{H}_2\text{O}$	+440
$\text{Fe}^{3+} + \text{e}^- \rightarrow \text{Fe}^{2+}$	+771
$\text{O}_2 + 4\text{H}^+ + 4\text{e}^- \rightarrow 2\text{H}_2\text{O}$	+840

Reakciók redoxpotenciálja



Potenciál a NADH/NAD arány függvényében



Microbes used in MFCs

Microbes	Substrate	Applications
<i>Actinobacillus succinogenes</i>	Glucose	Neutral red or thionin as electron mediator (Park and Zeikus, 2000; Park and Zeikus, 1999; Park et al., 1999)
<i>Aeromonas hydrophila</i>	Acetate	Mediator-less MFC Pham et al. (2003)
<i>Alcaligenes faecalis</i> , <i>Enterococcus gallinarum</i> , <i>Pseudomonas aeruginosa</i>	Glucose	Self-mediate consortia isolated from MFC with a maximal level of 4.31 W m ⁻² . Rabaey (2004)
<i>Clostridium beijerinckii</i>	Starch, glucose, lactate, molasses	Fermentative bacterium Niessen et al. (2004b)
<i>Clostridium butyricum</i>	Starch, glucose, lactate, molasses	Fermentative bacterium (Niessen et al., 2004b; Park et al., 2001)
<i>Desulfovibrio desulfuricans</i>	Sucrose	Sulphate/sulphide as mediator (Ieropoulos et al., 2005a; Park et al., 1997)
<i>Erwinia dissolven</i>	Glucose	Ferric chelate complex as mediators Vega and Fernandez, (1987)
<i>Escherichia coli</i>	Glucose sucrose	Mediators such as methylene blue needed. (Schroder et al., 2003; Ieropoulos et al., 2005a; Grzebyk and Pozniak, 2005)
<i>Geobacter metallireducens</i>	Acetate	Mediator-less MFC Min et al. (2005a)
<i>Geobacter sulfurreducens</i>	Acetate	Mediator-less MFC (Bond and Lovley, 2003; Bond et al., 2002)
<i>Gluconobacter oxydans</i>	Glucose	Mediator (HNQ, resazurin or thionine) needed Lee et al. (2002)
<i>Klebsiella pneumoniae</i>	Glucose	HNQ as mediator biomineralized manganese as electron acceptor (Rhoads et al., 2005; Menicucci et al., 2006)
<i>Lactobacillus plantarum</i>	Glucose	Ferric chelate complex as mediators (Vega and Fernandez, 1987)
<i>Proteus mirabilis</i>	Glucose	Thionin as mediator (Choi et al., 2003; Thurston et al., 1985)
<i>Pseudomonas aeruginosa</i>	Glucose	Pyocyanin and phenazine-1-carboxamide as mediator (Rabaey et al., 2004, 2005a)
<i>Rhodospirillum rubrum</i>	Glucose, xylose sucrose, maltose	Mediator-less MFC (Chaudhuri and Lovley, 2003; Liu et al., 2006)
<i>Shewanella oneidensis</i>	Lactate	Anthraquinone-2,6-disulfonate (AQDS) as mediator (Ringelisen et al., 2006)
<i>Shewanella putrefaciens</i>	Lactate, pyruvate, acetate, glucose	Mediator-less MFC (Kim et al., 1999a,b); but incorporating an electron mediator like Mn (IV) or NR into the anode enhanced the electricity production (Park and Zeikus, 2002)
<i>Streptococcus lactis</i>	Glucose	Ferric chelate complex as mediators (Vega and Fernandez, 1987)

Du et. al. 2007

