

**Particles/photons**

	I.	II.	III.
a	b		
p	e <sup>+</sup>	n	γ
α	e <sup>-</sup>		X

**Partners**

1. Electromagnetic field
2. Electron
3. Field of the nucleus
4. Nucleus

**Mechanism**

A) Absorption

B) Coherent scattering (only the direction is altered))

C) Incoherent scattering (also exchange of  $E$ )  
elastic (no excitation)  
inelastic

**Effect on radiation matter**

$\Delta I$   $E_{kin}, E^*$

$\Delta I$  -

$\Delta I, \Delta E$   
 $E_{kin}, E^*$  29

# 1. Ionizing radiations

## The first step of the ionizing radiation in the matter:

### 1. Neutral excitation



### 2. External ionization



### 3. Internal ionization



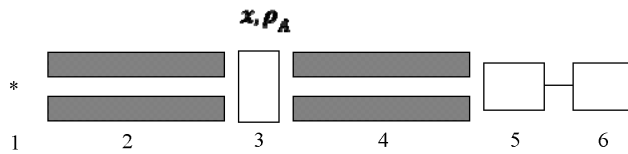
### 4. Bremsstrahlung (braking radiation)



FUNDAMENTALS OF DETECTION

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## Quantitative description of the interaction



$$v = \sigma n x \rho_A$$

cross section

$$-dn = \sigma(E)n\rho_A dx$$

$$n = n_0 e^{-\sigma(E)\rho_A x}$$

$$I = \frac{n}{t}$$

$$I = I_0 e^{-\mu x} \quad \text{linear absorption coefficient}$$

$$I = I_0 e^{-\mu x} = I_0 e^{-\frac{\mu}{\rho} x \cdot \rho} = I_0 e^{-\mu_m d} \quad \text{mass absorption coefficient}$$

$$x_{1/2} = \frac{\ln 2}{\mu} \quad d_{1/2} = \frac{\ln 2}{\mu_m}$$

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### $\alpha$ -radiation

Heavy, charged, high energy

With electrons: incoherent scattering

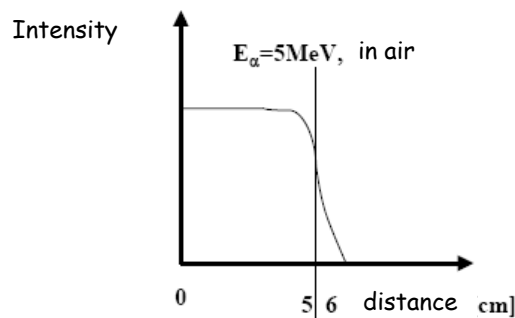
ionisation and excitation (50-50 %)

$E$  and direction of the alpha particles is modified

With the nucleus: Rutherford-scattering

nuclear reaction (see later)

**! Bremsstrahlung (continuous energy gamma radiation)!**



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### $\beta$ -radiation small, charged, limited energy

With electron: incoherent scattering

ionisation (external and internal)

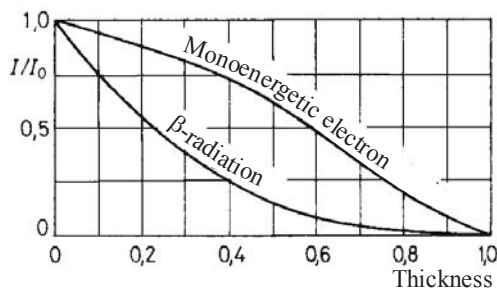
excitation

$E$  and the direction of the radiation changes

$$\left(\frac{dE}{dx}\right)_r = \frac{EZ}{800}$$
$$\left(\frac{dE}{dx}\right)_{\text{ion}}$$

With the field of the nucleus: incoherent scattering

**! Bremsstrahlung !**



$$I = I_0 e^{-\mu'x} = I_0 e^{-\mu d}$$

Linear/mass absorption coefficient <sup>34</sup>

Calculate the activity of 1 kg KCl. 0.012 % of the K atoms is radioactive  $^{40}\text{K}$ . The half life of  $^{40}\text{K}$  is  $1.13 \cdot 10^9$  years.

We prepared a  $^{35}\text{S}$  labelled protein at 12:00, 10 September 2014. The half life of the pure  $\beta^-$  emitter is 88 days. This sample was measured at noon on 26 September and the intensity was found 7000 imp/s. The overall efficiency of the measurement was 22 %. Calculate the activity of the sample in the time of synthesis.

The linear absorption coefficient of gamma radiation of 660 keV in aluminum is  $3,4 \text{ cm}^{-1}$ . Calculate the half thickness. How efficiently will attenuate this radiation an 10 cm aluminum wall ?

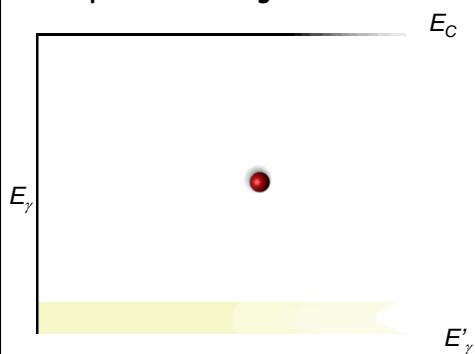
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## $\gamma$ -radiation

## electromagnetic radiation

### 1. Compton-scattering

Elastic collision of the photon with an electron

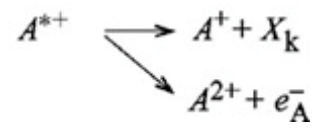
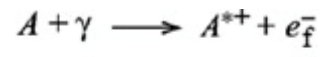
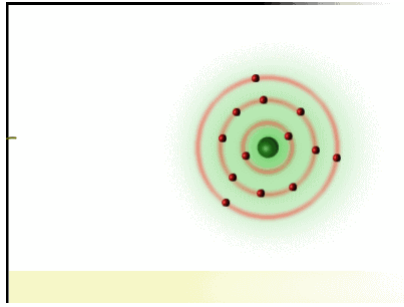


$$\mu_C = \frac{\mu'_C}{\rho} = \sigma_C \frac{\rho_A}{\rho} = \sigma_C \frac{N_A Z}{A}$$

$$\sigma_C = \sigma_s + \sigma_a$$

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## 2. Photoelectric effect

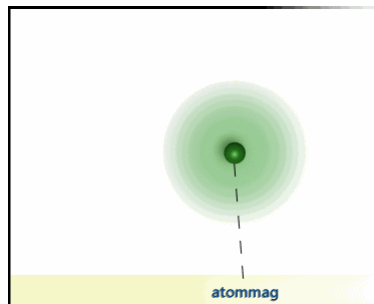


$$\sigma_{\text{f}} \approx \text{konst.} \frac{Z^n}{(\hbar \nu)^3}$$

$$n(E) = 4 - 5$$

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## 3. Pair production

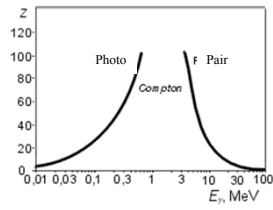
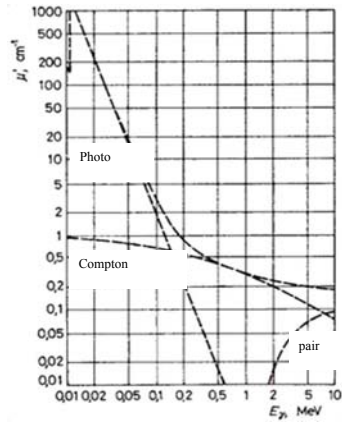


$$\sigma_{\text{p}} = K (E_{\gamma} - 1,02)^{2,2} Z^2$$

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$$I = I_0 e^{-\mu d} = I_0 e^{-(\mu_C + \mu_f + \mu_p)d}$$

### Germanium



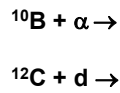
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## 2. Nuclear reactions

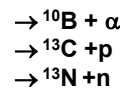
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## Cross section (~probability)

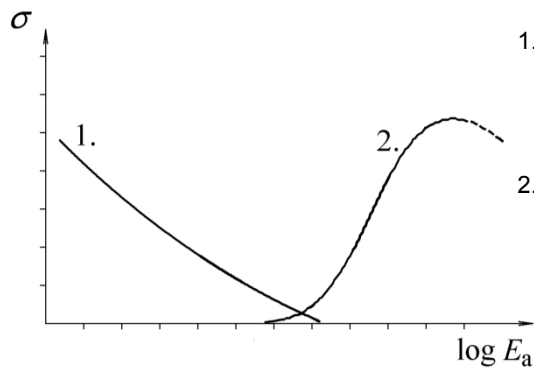
Conventional equation



$^{14}\text{N}^*$



Transition state



1. (n,γ)  
(n,f)  $^{233}\text{U}$ ,  $^{235}\text{U}$ ,  $^{239}\text{Pu}$ ,  $^{241}\text{Pu}$   
 $^{10}\text{B}(n,\alpha)$   
 $^6\text{Li}(n,\alpha)$
2. (γ,n)  
(n,2n)  
(n,α)  
(p, )  
(d, )

Tunnel effect

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## Kinetics of the nuclear reactions

$$\frac{dN^*}{dt} = \sigma_a N \phi - \lambda N^*$$

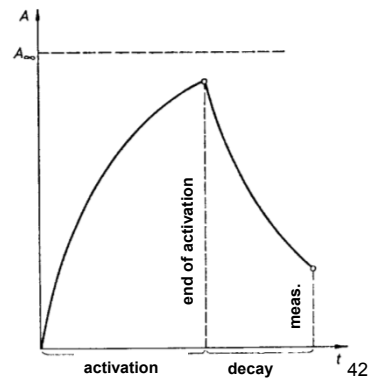
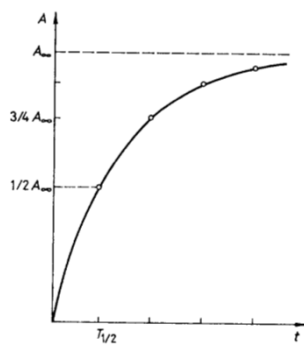
$$N^* = N_{\infty}^* [1 - \exp(-\lambda t)]$$

$$A = A_{\infty} [1 - \exp(-\lambda t)]$$

$$A_{\infty} = \lambda N_{\infty}^* = \phi \sigma_a N$$

$$A' = \lambda N^* =$$

$$= A_{\infty} [1 - \exp(-\lambda t)] \exp(-\lambda t_h)$$



We intend to obtain  $^{65}\text{Ni}$  with neutron irradiation. Therefore, we expose 1 g of Ni (with a  $^{64}\text{Ni}$  content of 91 %) to neutrons with a flux  $\Phi=10^{12}$  1/cm<sup>2</sup>s. The cross section  $\sigma$  of the



reaction is  $1.55 \cdot 10^{-28}$  m<sup>2</sup>. The half-life of  $^{65}\text{Ni}$  is 2.52 h.

- i) How long should the irradiation last if we want to reach 80 % of the saturation activity?
- ii) Estimate the ratio of the  $^{64}\text{Ni}/^{65}\text{Ni}$  isotopes in the sample after being „cooled“ for the same period as the activation lasted.